# organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

# Yan Yu, Chang Liu and Xingyu Yang\*

Department of Chemistry, Central China Normal University, Wuhan, Hubei 430079, People's Republic of China

Correspondence e-mail: yuyan801206@yahoo.com.cn

#### **Key indicators**

Single-crystal X-ray study T = 292 K Mean  $\sigma$ (C–C) = 0.003 Å R factor = 0.047 wR factor = 0.120 Data-to-parameter ratio = 15.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 5,5-Dimethyl-2-oxo-4-phenyl-2-(2-pyridyl-amino)-1,3,2-dioxaphosphorinane

In the crystal structure of the title compound,  $C_{16}H_{19}N_2O_3P$ , molecules are linked by two N-H···O hydrogen bonds, forming rings with an  $R_2^2(8)$  motif;  $\pi$ - $\pi$  stacking also contributes to the packing.

Received 27 October 2005 Accepted 8 November 2005 Online 26 November 2005

#### Comment

2-Chloro-1,3,2-dioxaphosphorinane is an important heterocycle, and its derivatives show good biological and pharmaceutical activity (Wolter & Hans, 1985; Jacobson & Nguyan, 1991; Rui *et al.*, 1997; Yang *et al.*, 1991); they also exhibit good fungicidal or antitumor activities. We report here the crystal structure of the title compound, (I).



The structure of (I) (Fig. 1) shows that the P atom of the oxodioxaphosphorinane ring carries a pyridylamine substituent, with a phenyl substituent at C7 and two methyl groups on C8. Overall, the oxaphosphorinane ring adopts a chair conformation. In the crystal structure,  $N-H\cdots$ O hydrogen bonds link neighboring molecules, forming inversion-related rings in an  $R_2^2(8)$  motif (Fig. 2) (Bernstein *et al.*, 1995). The distance between the inversion-related C1–C6 ring centroids  $(Cg\cdots Cg^i)$  is 3.5921 (13) Å [symmetry code: (i) 2 - x, -y, -z],



© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved

suggesting that significant  $\pi$ - $\pi$  interactions further stabilize the structure.

# **Experimental**

The title compound, (I), was prepared according to the procedure of Maier (1976); suitable crystals were obtained by vapor diffusion of dioxane into a dimethylformamide solution at room temperature (m.p. 515 K). IR (KBr,  $\nu$  cm<sup>-1</sup>): 3147, 1597, 1469, 1221, 1043, 1007, 971; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  7.33 (*s*, 5H), 8.30–6.90 (*m*, 4H), 9.09 (*s*, 1H), 4.45–3.95 (*dd*, 3H), 1.16 (*s*, 3H), 0.77 (*s*, 3H); analysis calculated for C<sub>16</sub>H<sub>19</sub>N<sub>2</sub>O<sub>3</sub>P: C 60.37 H 6.02, P 9.73%; found: C 60.21, H 6.12, P 9.60%.

#### Crystal data

$C_{16}H_{19}N_2O_3P$	$D_x = 1.335 \text{ Mg m}^{-3}$		
$M_r = 318.30$	Mo $K\alpha$ radiation		
Monoclinic, $P2_1/c$	Cell parameters from 3123		
$a = 10.3775 (9) \text{\AA}$	reflections		
b = 7.4406 (7)  Å	$\theta = 2.6-27.8^{\circ}$		
c = 20.6390 (18)  Å	$\mu = 0.19 \text{ mm}^{-1}$		
$\beta = 96.231 \ (2)^{\circ}$	T = 292 (2) K		
V = 1584.2 (2) Å <sup>3</sup>	Block, colorless		
Z = 4	$0.30 \times 0.20 \times 0.10 \ \text{mm}$		
Data collection			
Bruker SMART CCD area-detector	3107 independent reflections		
diffractometer	2913 reflections with $I > 2\sigma(I)$		

 $R_{\rm int} = 0.023$ 

 $\theta_{\rm max} = 26.0^{\circ}$ 

 $h = -12 \rightarrow 11$ 

 $k = -7 \rightarrow 9$ 

 $l = -25 \rightarrow 24$ 

 $w = 1/[\sigma^2(F_0^2) + (0.0529P)^2]$ 

where  $P = (F_0^2 + 2F_c^2)/3$ 

+ 0.8267P]

 $\Delta \rho_{\rm min} = -0.39 \ {\rm e} \ {\rm \AA}^{-3}$ 

 $(\Delta/\sigma)_{\rm max} = 0.001$  $\Delta\rho_{\rm max} = 0.22 \text{ e} \text{ Å}^{-3}$ 

 $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Bruker, 2000)  $T_{\min} = 0.946, T_{\max} = 0.982$ 23120 measured reflections

#### Refinement

Refinement on $F^2$	
$R[F^2 > 2\sigma(F^2)] = 0.047$	
$wR(F^2) = 0.120$	
S = 1.12	
3107 reflections	
204 parameters	
H atoms treated by a mixture of	
independent and constrained	
refinement	

## Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$\frac{\text{N1}-\text{H1}A\cdots\text{O3}^{\text{i}}}{\text{C10}-\text{H10}B\cdots\text{N2}}$	0.85 (1)	1.98 (1)	2.823 (2)	171 (2)
	0.96	2.57	3.522 (3)	174

Symmetry code: (i) -x + 1, -y, -z.



#### Figure 2

A view of the crystal packing. Hydrogen bonds are shown as dashed lines.

The coordinates of amino atom H1A were refined freely with  $U_{iso}(H) = 1.5U_{eq}(N)$ . All other H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C-H distances in the range 0.93–0.98 Å and  $U_{iso}(H) = 1.2U_{eq}(C)$  for aromatic or methylene and  $1.5U_{eq}(C)$  for methyl H atoms.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

The authors acknowledge financial support from the Science Foundation of Wuhan City (grant No. 20021002052).

### References

- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555–1573.
- Bruker (1997). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2000). *SMART*, *SAINT* and *SADABS* (Version 6.10). Bruker AXS Inc., Madison, Wisconsin, USA.
- Jacobson, R. M. & Nguyan, L. Tu. (1991). Patent Appl. EP 437 335.
- Maier, L. (1976). Metal Org. Chem. 6, 133-155.
- Rui, L. S., Guang, F. Y. & Wei, S. M. (1997). Chin. Chem. Lett. 8, 855-858.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Wolter, T. H. & Hans, W. (1985). J. Org. Chem. Lett. 50, 4508-4514.
- Yang, H. Z, Wu, Y. & Zhang, Y. F. (1991). Chem. J. Chin. Univ. 12, 44.